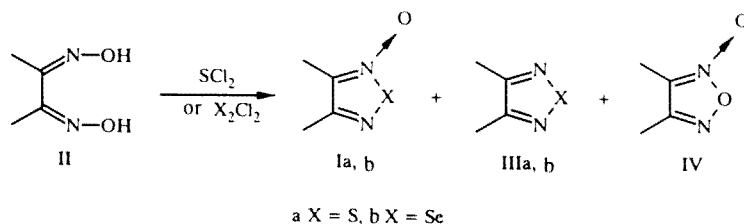


NEW EXAMPLES OF THE USE OF SULFUR AND SELENIUM MONOCHLORIDES IN THE SYNTHESIS OF AZOLES

A. A. Yavolovskii, E. A. Kuklenko and É. I. Ivanov

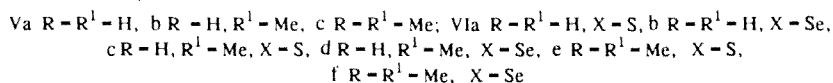
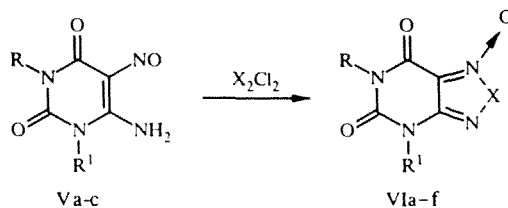
Sulfur and selenium monochlorides can be used in the synthesis of new heterocyclic systems: 1,2,5-thiadiazolo- and 1,2,5-selenodiazolo-[3,4-d]pyrimidine-5,7-(4H,7H)dione N-oxides and isothiazolo- and isoselenoazolof[4,3-e]-5,6-dihydro-4H-1,4-diazepines.

Although 1,2,5-oxadiazolo-N-oxides (furoxanes) have been known for a long time and have been studied extensively, the first report on the structure of 1,2,5-thiadiazole N-oxides (Ia) only appeared in 1970 [1]. 1,2,5-Thiadiazole N-oxides were obtained in low yield by the reaction of 1,2-diketone dioximes (II) with an excess of sulfur dichloride in benzene or sulfur monochloride in DMF [2]; the main reaction products were the corresponding 1,2,5-thiadiazoles (IIIa) [1]:



Among the products from the reaction of dioxime II with selenium monochloride [3, 4], apart from the N-oxide (Ib), were considerable amounts of the selenodiazole (IIIb) and the furoxane (IV). Unfortunately, there has been no further study of this interesting class of heterocycle.

We have found that 1,2,5-thiadiazole and 1,2,5-selenodiazole N-oxides can be obtained from heterocyclic *o*-nitrosoamines with the help of sulfur and selenium monochlorides. We have used as model compounds the 5-nitroso-6-aminouracils (Va-c), which are widely used in the Traube synthesis of purines [5-7]:



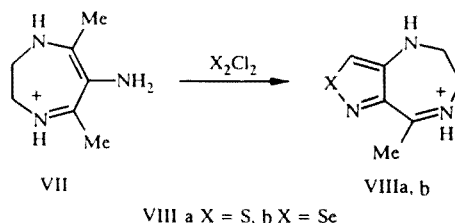
Treatment of Va-c with sulfur or selenium monochlorides in DMF at temperatures of 40-60°C gave the 1,2,5-thiadiazolo- or 1,2,5-selenodiazolo[3,4-d]pyrimidin-5,7-(4H,6H) dione N-oxides (VIa-f), which are high-melting (with decomposition) compounds which dissolve poorly in water or organic solvents. The most characteristic feature of the mass spectra of these compounds is that the molecular ion has the highest intensity. Other features are the loss of oxygen from the molecular

A. V. Bogatskii Physicochemical Institute, National National Academy of Sciences of Ukraine, Odessa 270080. Translated from *Khimiya Geterotsiklicheskikh Soedinenii*, No. 7, pp. 997-999, July, 1996. Original article submitted May 27, 1996.

TABLE 1. Characteristics of the Compounds Synthesized

Compound	Molecular formula	Found, %			Calculated, %		
		C	H	N	C	H	N
VIa	C ₄ H ₂ N ₄ O ₃ S	25,70	1,10	30,18	25,81	1,08	30,11
VIb	C ₄ H ₂ N ₄ O ₃ Se	20,41	0,83	24,1	25,51	0,85	23,93
VIc	C ₅ H ₄ N ₄ O ₃ S	30,16	1,84	27,80	30,00	2,00	28,00
VI d	C ₅ H ₄ N ₄ O ₃ Se	23,92	1,59	22,61	24,19	1,61	22,58
VIe	C ₆ H ₆ N ₄ O ₃ S	33,59	2,78	26,20	33,64	2,80	26,17
VI f	C ₆ H ₆ N ₄ O ₃ Se	27,42	2,34	21,35	27,48	2,29	21,37
VIIIa	C ₇ H ₉ N ₃ S	50,19	5,40	25,10	50,30	5,39	25,15
VIIIb	C ₇ H ₉ N ₃ Se	39,00	4,24	19,56	39,07	4,19	19,53

ion radical and the elimination of RNCO. The IR spectra show intense N–O stretching frequencies at 1260-1220 cm⁻¹. We have obtained 1,2-thiazoles and 1,2-selenodiazoles from 6-amino-2,3-dihydro-5,7-dimethyl-1,4-diazepin (VII) [8] using sulfur and selenium monochlorides:



Annelated isothiazoles and isoselenoazoles have been obtained previously by the reaction of aromatic or heteroaromatic *o*-methylanilines with *N*-sulfinylmethanesulfonamides [9, 11] or thionyl chloride and selenious acid in dioxane [12].

We also obtained compound VIIIa by treatment of VII with thionyl chloride in THF but no positive results were obtained when selenious acid was used in an attempt to make VIIIb. The mass spectra of compounds VIII a and b contain intense molecular ions. The ¹H NMR spectra are characterized by signals of the aromatic proton of the isothiazolo and isoselenoazolo rings at 7.90 and 8.20 ppm respectively, by multiplets of the ethylene fragment of the diazepin ring at 3.30 ppm (VIIIa) and 3.50 ppm (VIIIb), and by singlets of the exocyclic methyl groups at 1.90 ppm (VIIIa) and 2.40 ppm (VIIIb).

EXPERIMENTAL

Purity of the products was monitored by TLC on Silufol UV 254-vis strips. Mass spectra were recorded with a Varian MAT-112 with direct inlet and an ionizing energy of 70 eV. IR spectra of thin films in Nujol were recorded with a Specord-80 spectrometer, and ¹H NMR spectra of CF₃CO₂D solutions were recorded with a Tesla BS-497 (100 MHz) machine.

1,2,5-Thiadiazolo[3,4-*d*]pyrimidin-5,7-(4H,6H)dione N-oxides (VI: a — C₄H₂N₄O₃S; c — C₅H₄N₄O₃S; e — C₆H₆N₄O₃S). Sulfur monochloride (10 cm³, 0.125 mole) was added with intense stirring to a solution of a 5-nitroso-6-aminouracil (0.1 mole) in dry DMF (50 cm³) at such a rate that the temperature did not exceed 60°C. The precipitate was filtered off and purified either by precipitation from basic solution with hydrochloric acid or by recrystallization from DMF to give VIa (7.9 g, 43%), mp > 300°C, M⁺ 186, IR spectrum 3560-3000, 1720, 1650, 1610, 1520, 1260 cm⁻¹, or VIc (15 g, 75%), mp > 300°C, M⁺ 200, IR spectrum 3140-3000, 1730, 1690, 1530, 1510, 1240 cm⁻¹, or VIe (10 g, 47%), mp > 300°C, M⁺ 214, IR spectrum 1700, 1660, 1520, 1240 cm⁻¹.

1,2,5-Selenodiazolo[3,4-*d*]pyrimidin-5,7-(4H,6H)dione N-oxides (VI: b — C₄H₂N₄O₃Se; d — C₅H₄N₄O₃Se; f — C₆H₆N₄O₃Se). Selenium monochloride (10 cm³, 0.125 mole) was added with intense stirring to a solution of a 5-nitroso-6-aminouracil (0.1 mole) in dry DMF (50 cm³) at such a rate that the temperature did not exceed 40°C. The precipitate was recrystallized from DMF to give VIb (8.6 g, 37%), mp > 300°C, M⁺ 234, IR spectrum 3550-3000, 1700, 1600, 1560, 1500, 1240 cm⁻¹, or VI d (8.0 g, 32%), mp > 300°C, M⁺ 248, IR spectrum 3140-3000, 1710, 1680, 1510, 1230 cm⁻¹, or VI f (10 g, 38%), mp > 300°C, M⁺ 262, IR spectrum 1690, 1650, 1520, 1230 cm⁻¹.

8-Methylisothiazolo[4,3-*e*]-5,6-dihydro(4H)-1,4-diazepin Perchlorate (VIIIa, C₇H₉N₃S·HClO₄). Sulfur monochloride (2 cm³, 0.02 mole) was added to 6-amino-2,3-dihydro-5,7-dimethyl(1H)-1,4-diazepin (VIII) (2.39 g, 0.01 mole) in dry

DMF (50 cm³). The mixture heated up spontaneously. The solvent was decanted from the oily product which was dissolved in hot ethanol. The solution was boiled with charcoal and filtered. The crystals which separated were recrystallized from 2:3 ethanol–acetone to give VIIIa (1.70 g, 64%), mp 270°C, M⁺ 167 (for the free base). ¹H NMR spectrum: 1.9 (3H, s, CH₃), 3.0 (4H, m, C₂H₄), 7.9 ppm (1H, s, 3-H). IR spectrum: 3185-3100, 1730, 1700, 1670, 1600 and 1560 cm⁻¹.

8-Methyliselenoazolo[4,3-*e*]-5,6-dihydro(4H)-1,4-diazepin perchlorate (VIIIb, C₇H₉N₃Se·HClO₄) was prepared analogously to VIIIa (yield 34%), mp 275°C, M⁺ 215 (for the free base). ¹H NMR spectrum: 2.4 (3H, s, CH₃), 3.5 (4H, m, C₂H₄), 8.2 (1H, s, 3-H). IR spectrum: 3230-3100, 1670, 1610, 1580, 1550 and 1510 cm⁻¹.

REFERENCES

1. K. Pilgram, *J. Org. Chem.*, **35**, 1165 (1970).
2. V. G. Pesin, A. M. Khaletskii, and Chou-Chin, *Zh. Org. Khim.*, **28**, 2131 (1958).
3. C. L. Pederson, *J. Chem. Soc. Chem. Commun.*, No. 4, 704 (1974).
4. C. L. Pederson, *Acta Chem. Scand.*, **30**, 675 (1976).
5. W. Traube, *Ber.*, **33**, 3035 (1900).
6. J. A. Lister, *Fused Pyrimidines*, Pt. 2. Purines, J. Brown and S. C. Juter (eds.), Wiley, New York (1971).
7. N. V. Rubtsov and A. G. Baichikov, *Synthetic Chemico-Pharmaceutical Materials* [in Russian], *Meditina*, Moscow (1971), p. 283.
8. P. Lloyd, R. H. McDougall, and D. R. Marshall, *J. Chem. Soc. (C)*, No. 5, 780 (1966).
9. B. Danylec and M. Davis, *J. Heterocycl. Chem.*, **17**, 533 (1980).
10. B. Danylec and M. Davis, *J. Heterocycl. Chem.*, **17**, 537 (1980).
11. C. M. Singerman, *J. Heterocycl. Chem.*, **12**, 877 (1975).
12. T. Ueda, H. Yoshida, and J. Sakakibara, *Synthesis*, Nos. 6-7, 695 (1985).